

# Single crystalline SnO<sub>2</sub> thin films grown on *m*-plane sapphire substrate by mist chemical vapor deposition

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Tin dioxide (SnO<sub>2</sub>) thin films, as a candidate for realizing next-generation electrical and optical devices, were grown on 2-inch diameter *m*-plane sapphire substrates by mist chemical vapour deposition at atmospheric pressure. The SnO<sub>2</sub> thin films were characterized by scanning electron microscope (SEM), atomic force microscope (AFM), X-ray diffraction (XRD) in  $\theta$ - $2\theta$  and  $\phi$  scanning modes, and electron backscatter diffraction (EBSD). Although the SEM and AFM images showed a relatively rough surface morphology, it was found from the XRD and EBSD measurements that SnO<sub>2</sub> films were epitaxially grown on the

substrates under optimised growth condition. Epitaxial growth of SnO<sub>2</sub> thin film growth at three typical areas on the substrate was confirmed by the EBSD measurements. It is likely that the single crystalline SnO<sub>2</sub> (001) thin film was formed across the 2-inch sapphire substrate. Finally, the second SnO<sub>2</sub> layer was overgrown on the above single crystalline SnO<sub>2</sub> thin film, which functioned as a buffer layer. This method which drastically improved surface roughness of the second SnO<sub>2</sub> layer.

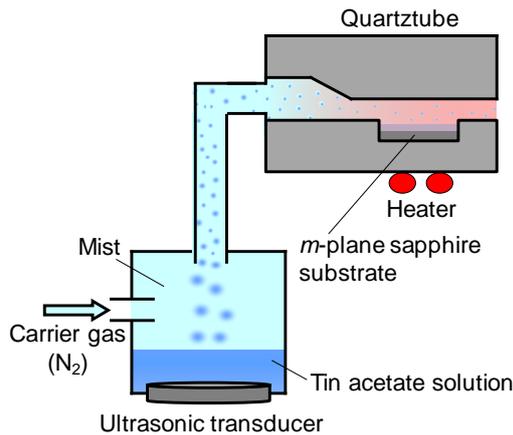
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**1 Introduction** Oxide semiconductors have recently attracted much attention due to their potential use in the next-generation electronics devices such as thin-film transistor [1], light-emitting diodes [2], solar cells [3], power devices [4] and so on. Tin dioxide (SnO<sub>2</sub>) is a direct forbidden band gap [5] oxide semiconductor with unique intrinsic material properties, which include wide band gap ( $\sim 4$  eV) [6-10], high exciton binding energy ( $\sim 130$  meV) and high carrier mobility ( $\sim 250$  cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup>) at room temperature [11]. In addition, SnO<sub>2</sub> is a relatively low-cost material because it does not contain minor metals such as gallium. In order to obtain high crystallographic properties of SnO<sub>2</sub>, the growth of single crystalline SnO<sub>2</sub> require relatively expensive film formation methods such as metal-organic chemical vapour

deposition (MOCVD) [10] or molecular beam epitaxy (MBE) [12,13], which needs vacuum equipment. One alternative approach is a mist chemical vapour deposition (mist-CVD), which can form various oxide semiconductors under atmospheric pressure with a simple and less expensive technique [9,14-19].

Most of the previous studies targeted SnO<sub>2</sub> film grown on *c*-plane and *r*-plane sapphire substrate even under vacuum condition (MOCVD, MBE and so on) and only few studies have focused on the film grown on other planes of sapphire [10]. It is clear that lattice constants of sapphire substrate and SnO<sub>2</sub> film should match as close as possible for hetero-epitaxial deposition to be successful. Since the lattice mismatch (9.5 %) between SnO<sub>2</sub> and the *m*-plane

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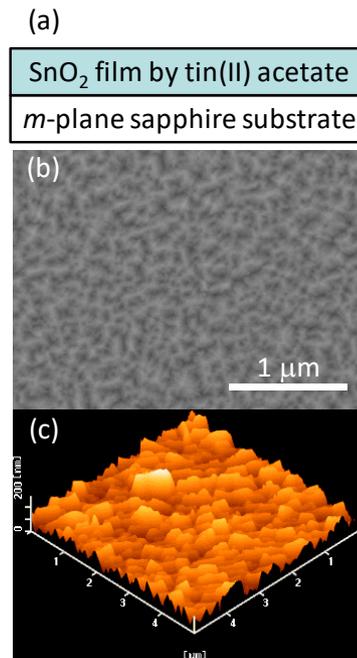
**Figure 1** Schematic illustration of the fine-channel type mist-CVD system used in this study.

sapphire along  $\text{SnO}_2$  [010] is smaller than that between  $\text{SnO}_2$  and  $c$ -plane (16.0 %),  $\text{SnO}_2$  and  $r$ -plane sapphire (11.3 %) [20],  $\text{SnO}_2$  film is favorable to grow on  $m$ -plane sapphire substrate.

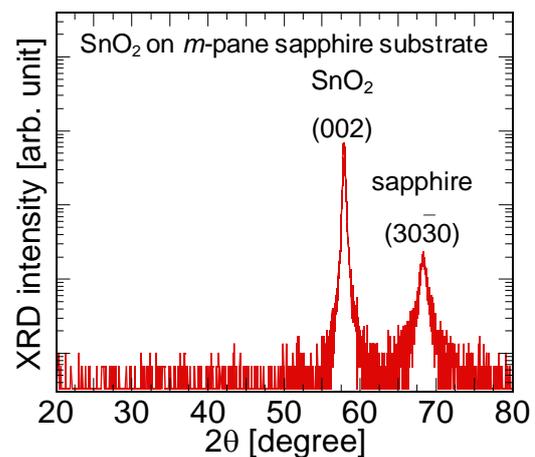
In the present work,  $\text{SnO}_2$  films grown on  $m$ -plane sapphire substrates by mist-CVD involve both low-cost materials and procedures, because Sn and O are abundant, and mist-CVD is a simple atmospheric-pressure process. The detailed structural properties of the films were investigated.

**2 Experiments** In this study,  $\text{SnO}_2$  films were grown using a homemade fine-channel-type mist-CVD system, which was operated at atmospheric pressure. Figure 1 shows a schematic illustration of the mist-CVD system used in this study. The source of the mist was tin(II) acetate  $((\text{CH}_3\text{COO})_2\text{Sn})$  solution with concentration of 0.04 mol/L. The source solution was ultrasonically atomized at the frequency of 2.4 MHz, and the mist particles were transferred to the reaction area with nitrogen carrier gas with a flow rate of 4 L/min.  $\text{SnO}_2$  films were grown on 2-inch diameter  $m$ -plane sapphire substrates at 525 °C for 40 min without using any buffer layer. To estimate the thickness of the  $\text{SnO}_2$  film, we performed X-ray fluorescence (XRF) measurement at three positions on the sample. The thicknesses of  $\text{SnO}_2$  film at three positions on the sample were estimated to be 121, 108 and 114 nm, respectively. Their deviation was 6% from the average (114 nm).

**3 Results and discussion** We initially performed scanning electron microscope (SEM) and atomic force microscope (AFM) observation to characterize the  $\text{SnO}_2$  surface morphology. A top view SEM and AFM images of the  $\text{SnO}_2$  surface are shown in Fig. 2. The AFM observation confirmed that the root mean square (RMS) value of a  $5 \times 5 \mu\text{m}^2$  region is 29 nm. Although the SEM and AFM images showed a relative rough surface morphology as shown Fig.



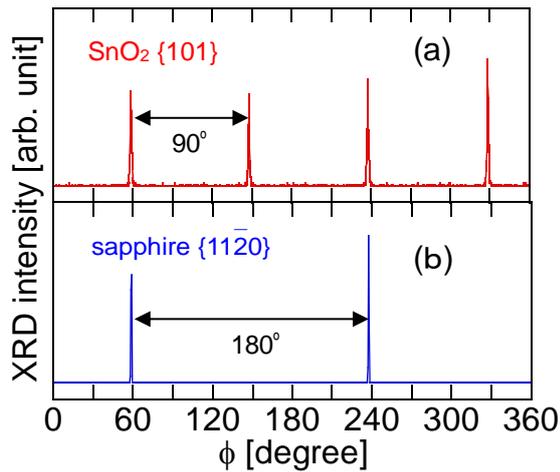
**Figure 2** (a) Schematic illustration of the sample structure, (b) top view SEM and (c) AFM images of the  $\text{SnO}_2$  film grown by mist-CVD with tin(II) acetate solution.



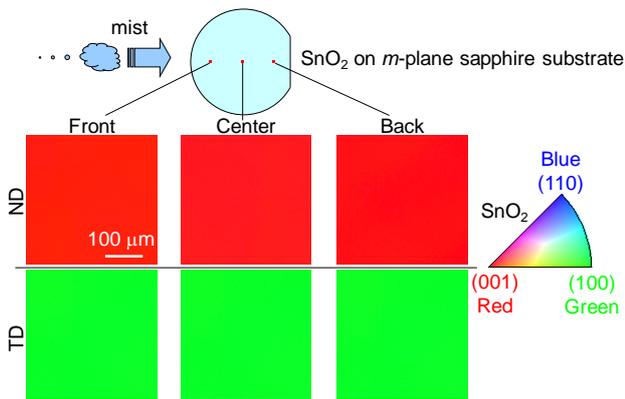
**Figure 3** Result of XRD  $\theta$ -2 $\theta$  scan for the  $\text{SnO}_2$  film grown on  $m$ -plane sapphire substrate.

2, the  $\text{SnO}_2$  surface morphology was much smoother than that of our previous result using the tin(II) chloride aqueous solution (not shown here). The reason for the obtained relatively flat surface morphology formed tin(II) acetate is not clear yet, though tin(II) chloride ( $\text{SnCl}_2$ ) is often used as a source solution to grow  $\text{SnO}_2$  by mist-CVD [9,18].

To characterize crystallographic properties of the  $\text{SnO}_2$  thin film, we carried out X-ray diffraction (XRD) in  $\theta$ -2 $\theta$  and  $\phi$  scanning modes. Figure 3 shows the XRD  $\theta$ -2 $\theta$  scan patterns measured using incident  $\text{Cu K}\alpha_1$  radiation. Only the  $\text{SnO}_2$  (002) and the sapphire  $m$ -plane of (3030)

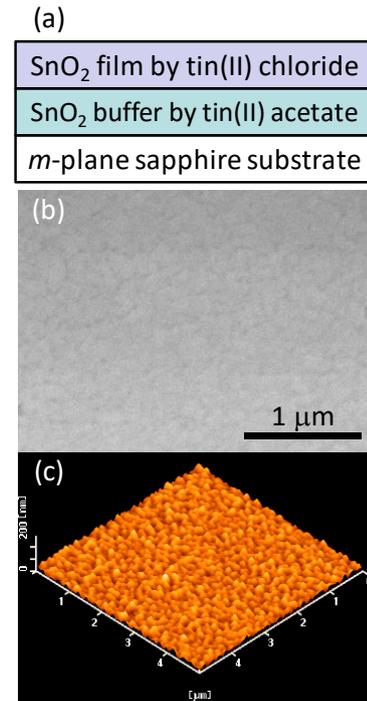


**Figure 4** XRD  $\phi$  scan for the (a) SnO<sub>2</sub> film and sapphire substrate.



**Figure 5** Both normal direction (ND) and transverse direction (TD) of EBSD images at different positions in the SnO<sub>2</sub> film.

were observed in the pattern, indicating that the SnO<sub>2</sub> film was (001) orientated. The XRD  $\phi$  scan was then measured to identify the lateral orientation of the SnO<sub>2</sub> film. The SnO<sub>2</sub> {101} and sapphire {1120} planes were used as the monitoring planes. Figure 4 shows the XRD  $\phi$  scan results. Four peaks spaced by 90° were observed for the SnO<sub>2</sub>, indicating the fourfold symmetry along [001] axis of SnO<sub>2</sub> with rutile structure and their epitaxial relationship of SnO<sub>2</sub> (001) [100] || sapphire (1010) [1210] [10]. These  $\theta$ -2 $\theta$  and  $\phi$  scan XRD results indicate that the SnO<sub>2</sub> (001) films were epitaxially grown on the sapphire (1010) substrates. Note that there was no significant difference among these XRD results at three locations on the sample, which indicate that uniform SnO<sub>2</sub> films were formed. To further investigate the epitaxial growth of the SnO<sub>2</sub> film, electron backscatter diffraction (EBSD) images were analysed to identify the crystallographic orientation at various locations on the sample. The spacing between neighbouring electron beams was 3  $\mu$ m. Figure 5 shows both normal direction (ND) and transverse direction (TD) of EBSD images of the SnO<sub>2</sub> film.



**Figure 6** (a) Schematic illustration of the sample structure, (b) top view SEM and (c) AFM images of the SnO<sub>2</sub> film with the buffer layer.

The similarity of the three EBSD images in Fig. 5 indicated that the crystallographic orientation of SnO<sub>2</sub> on the substrate was homogeneous. These results suggest that the single crystalline SnO<sub>2</sub> (001) thin film was formed across the 2-inch diameter *m*-plane (1010) sapphire substrate.

Even though the surface was not smooth in the above experiment because no buffer layer was used, surface roughness could be potentially improved using a buffer layer. It is a common knowledge that a hetero-epitaxial semiconductor layers need to be deposited with suitable buffer layers to preserve the high crystalline quality of the over layers [21]. Based on this idea, the single crystalline SnO<sub>2</sub> (001) thin film formed with above condition on the substrate was used as the buffer layer. Second layer of SnO<sub>2</sub> films were overgrown on the single crystalline SnO<sub>2</sub> buffer layer at 800 °C for 40 min. The source of the mist was changed to tin(II) chloride aqueous solution [9,18] with concentration of 2.5 mol/L. Figure 6 shows a top view SEM and AFM images of the surface of the second SnO<sub>2</sub> layer grown on the buffer layer. The AFM observation confirmed that the RMS value of a 5 × 5  $\mu$ m<sup>2</sup> region is 9 nm. The SEM and AFM measurements clearly indicated that the surface roughness was drastically improved using the buffer layer, and the detailed characterization will be continued.

**4 Conclusion** SnO<sub>2</sub> thin films were grown on 2-inch diameter *m*-plane sapphire substrates by mist-CVD at atmospheric pressure. The SnO<sub>2</sub> thin films were characterized by

SEM, XRD and EBSD. It was found from the XRD and EBSD measurements that SnO<sub>2</sub> films were epitaxially grown on the substrates under optimised growth condition. Epitaxial growth of SnO<sub>2</sub> thin film growth at three typical areas on the substrate was confirmed by the EBSD measurements. It is likely that the single crystalline SnO<sub>2</sub> (001) thin film was formed across the 2-inch sapphire substrate. Finally, the second SnO<sub>2</sub> layer was overgrown on the above single crystalline SnO<sub>2</sub> thin film, which functioned as a buffer layer. This method which drastically improved surface roughness of the second SnO<sub>2</sub> layer.

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