

Manganese(III)-Based Oxidative Dual Cyclization of 4,4-Diaryl-3-butenyl-3-oxobutanoates

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Abstract—The oxidative intramolecular tandem cyclization of 4,4-diaryl-3-butenyl 3-oxobutanoates **3a-e** and 4,4-diaryl-3-butenyl 2-(2-aryl-2-oxoethyl)-3-oxobutanoates **5aa-ee** using manganese(III) acetate was investigated. The reaction of oxobutanoates **3a-e** gave the desired dual cyclization products, 7,7-diaryl-9-methyl-3,8-dioxabicyclo[4.3.0]non-9-en-2-ones **4a-e**, in excellent yields. A similar cyclization of oxobutanoates **5aa-ee** did not give the tricyclic compounds **7** but 1-acetyl-7,7,9-triaryl-1,3,8-dioxabicyclo[4.4.0]dec-9-en-2-one **6aa-ee** in good yields.

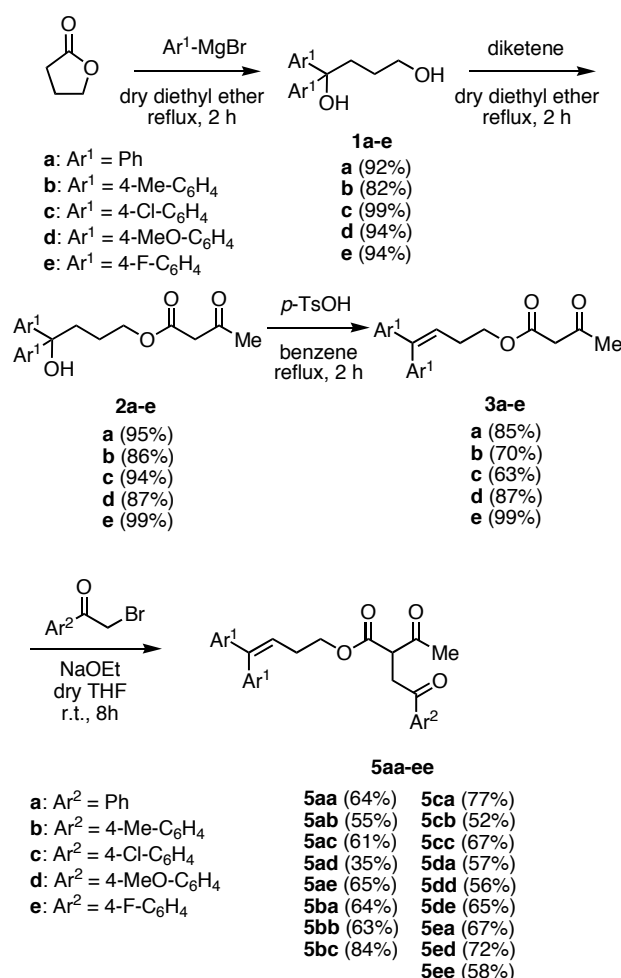
1. Introduction

One of the synthetic targets in the organic synthesis is heterocyclic compounds which have chemically interesting structure and useful actions in human life.¹ There are many naturally occurring oxygen-containing heterocyclic compounds, some of which have very strong biological activities.² For example, the trioxabicyclo[4.3.0]non-3-ene framework is also found in artemisinin, yingzhausu A and C, and etc.³ Artemisinin and yingzhausu A and C have an antimalarial activity. Moreover, these heterocyclic compounds are important as a building block that is essential for the total synthesis of many natural products.¹ For these reasons, to construct the dioxabicyclo[4.3.0]non-9-en-2-one and dioxabicyclo[4.4.0]dec-9-en-2-one skeleton is very significant in organic chemistry.^{4,6}

Recently, we reported the synthesis of dioxabicyclo[3.3.0]oct-3-enes and dihydropyranes using manganese(III)-based tandem cyclization of malonates or pentanediones with alkenes.⁷ In the course of our study, we planned to investigate a similar dual and/or three-dimensional cyclization using 4,4-diaryl-3-butenyl 3-oxobutanoates.^{8,9,10}

2. Results and Discussion

4,4-Diaryl-3-butenyl 3-oxobutanoates **3a-e** were prepared by the reaction of γ -butyrolactone with arylmagnesium bromides and then, the monoesterification of the obtained diols **1a-e** with diketene followed by dehydration. 4,4-Diaryl-3-butenyl 2-(2-aryl-2-oxoethyl)-3-oxobutanoates **5aa-ee** were synthesized by the reaction of 4,4-diaryl-3-butenyl 3-oxobutanoates **3a-e** with α -bromoacetophenones which were prepared by the reaction of the corresponding acetophenones with bromine in methanol (Scheme 1).

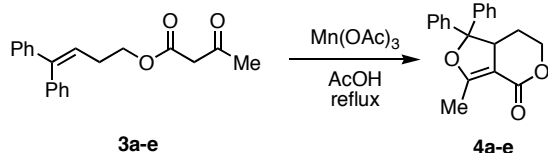


Scheme 1.

Oxidation of 4,4-diphenyl-3-butenyl 3-oxobutanoate (**3a**) with manganese(III) acetate in boiling acetic acid gave 9-methyl-7,7-diphenyl-3,8-dioxabicyclo[4.3.0]non-9-en-2-one (**4a**) in a 76% yield (Scheme 2 and Table 1, Entry 1). The reaction was scrutinized under various conditions and optimized the molar ratios, amounts of the solvent, reaction temperatures, and atmosphere. A similar reaction of other 4,4-diaryl-3-butenyl 3-oxobutanoates **3b-**

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e was carried out under the optimized reaction conditions to give similar bicyclic compounds **4b-e** in moderate to good yields (Table 1, Entries 2-5). The best yield was achieved by the reaction using electron-rich butanoate **3d**, and the corresponding bicyclic compound **4d** was obtained in a 95% yield. Probably, the dual cyclization would be accelerated during the reaction since the tertiary carbocation intermediate derived from **3d** would be stabilized by the electron-donating 4-methoxyphenyl group. As a result, the considerable substituent effect of butanoates **3a-e** was observed.



Scheme 2.

Table 1. Manganese(III)-Based Dual Cyclization of **3a-e**

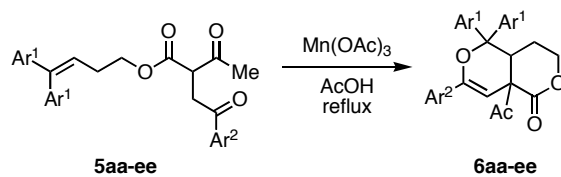
Entry	3:Mn(III) ^a	Time/min	Yield/% ^b
1	1:4	4	4a (76)
2	1:4	4	4b (82)
3	1:4	4	4c (72)
4	1:4	4	4d (95)
5	1:4	8	4e (62)

^a Molar ratio. The butanoate **3** (1.0 mmol) was used in boiling AcOH (50 mL).

^b Isolated yield based on the butanoate **3** used.

In order to further investigate the tandem cyclization, the reaction of 4,4-diaryl-3-butenyl 2-(2-aryl-2-oxoethyl)-3-oxobutanoates **5aa-ee** was examined (Scheme 3). When the oxidation of 4,4-diphenyl-3-butenyl 2-(2-oxo-2-phenylethyl)-3-oxobutanoate (**5aa**) with manganese(III) acetate was carried out in boiling acetic acid, 1-acetyl-7,7,9-triphenyl-3,8-dioxabicyclo[4.4.0]dec-9-en-2-one (**6aa**) was obtained in a 57% yield (Table 2, Entry 1). The reaction was scrutinized under various reaction conditions and optimized the molar ratios, amounts of the solvent, reaction temperatures, and atmosphere. Unfortunately, the desired three-dimensional product **7** was not isolated. A similar oxidation of other 4,4-diaryl-3-butenyl 2-(2-aryl-2-oxoethyl)-3-oxobutanoates **5** with manganese(III) acetate was conducted under the optimized reaction conditions, giving similar bicyclic compounds **6** in moderate to good yields (Table 2, Entries 2-17). In this case, a similar substituent effect for the formation of **6** was observed as well as that for the production of dioxabicyclic compounds **4**.

The mechanism for the manganese(III)-based tandem cyclization of oxobutanoates **5** was outlined in Scheme 4.



Scheme 3.

Table 2. Manganese(III)-Based Dual Cyclization of **5aa-ee**

Entry	5:Mn(III) ^a	Time/min	Yield of 6 (%) ^b
1	1:04	5	6aa (57)
2	1:03	3	6ab (63)
3	1:03	5	6ac (43)
4	1:04	4	6ad (65)
5	1:04	10	6ae (35)
6	1:03	2	6ba (58)
7	1:04	4	6bb (67)
8	1:04	8	6bc (60)
9	1:03	4	6ca (43)
10	1:04	8	6cb (55)
11	1:03	5	6cc (41)
12	1:04	3	6da (51)
13	1:03	1	6dd (68)
14	1:03	2	6de (39)
15	1:04	7	6ea (42)
16	1:03	4	6ed (58)
17	1:04	10	6ee (26)

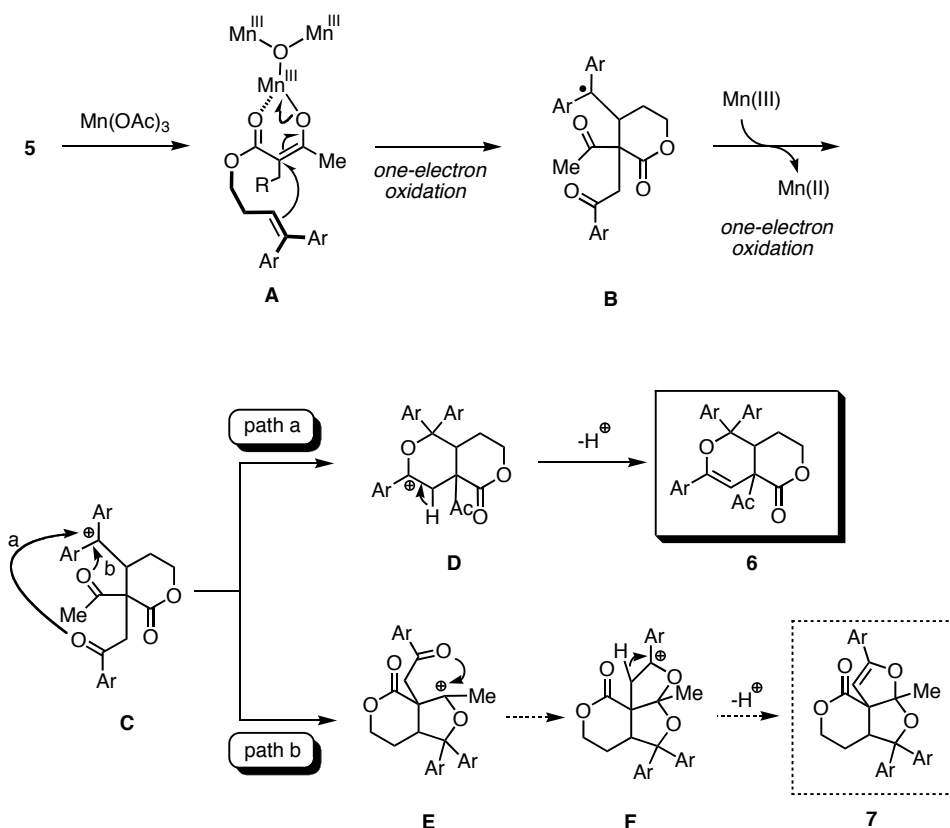
^a Molar ratio. The butanoate **5** (0.50 mmol) was used in boiling AcOH (20 mL).

^b Isolated yield based on the butanoate **5** used.

Although the dual cyclization of oxobutanoates **3** and **5** exactly occurred (path a), the 2-aryl-2-oxoethyl-substituted oxobutanoates **5** did not undergo the three-dimensional cyclization (path b).

3. Conclusion

The reaction of 4,4-diaryl-3-butenyl 3-oxobutanoates **3a-e** using manganese(III) acetate was investigated in detail. As a result, the reaction in acetic acid at reflux temperature gave dioxabicyclo[4.3.0]non-9-en-2-ones **4a-e**. On the other hand, a similar reaction of 4,4-diaryl-3-butenyl 2-(2-aryl-2-oxoethyl)-3-oxobutanoates **5aa-ee** was carried out using manganese(III) acetate in acetic acid at reflux temperature gave dioxabicyclo[4.4.0]dec-9-en-2-ones **6aa-ee**. The mechanism for the formation of the products **4** and **6** was explained based on the tandem cyclization pathway.



Scheme 4.

Acknowledgments

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